

## UČNI NAČRT PREDMETA / COURSE SYLLABUS

<b>Predmet:</b>	Trajnostni vidiki v procesni tehniki
<b>Course title:</b>	Sustainable Process Engineering

Študijski program in stopnja Study programme and level	Študijska smer Study field	Letnik Academic year	Semester Semester
Sonaravne tehnologije in sistemi v strojništvu - 3. stopnja	/	1./2.	zimski/letni
Sustainable technologies and systems in mechanical engineering - 3 <sup>rd</sup> cycle	/	first/second	winter/summer

<b>Vrsta predmeta / Course type</b>	izbirni/elective
-------------------------------------	------------------

<b>Univerzitetna koda predmeta / University course code:</b>	
--	--

Predavanja Lectures	Seminar Seminar	Vaje Tutorial	Laboratorijske vaje work	Druge oblike študija	Samost. delo Individ. work	ECTS
10		30	10	/	250	10

<b>Nosilec predmeta / Lecturer:</b>	izr. prof. dr. Viktor Grilc
-------------------------------------	-----------------------------

<b>Jeziki / Languages:</b>	<b>Predavanja / Lectures:</b> slovenski/Slovenian
	<b>Vaje / Tutorial:</b> slovenski/Slovenian

**Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:**

- Izpolnjeni splošni pogoji za vpis na ta nivo študija.

Fulfilled general entrance qualifications for this level of studies

### Vsebina:

- Osnovni pojmi in definicije procesne tehnike.
- Prenos toplote. Mehanizmi prenosa toplote. Stacionarno prevajanje. Nestacionarno prevajanje. Konvektivni prenos toplote. Koeficienti toplotnega prestopa. Korelacije.
- Prenos snovi. Mehanizmi snovnega prenosa. Stacionarna difuzija. Konvektivni snovni prenos. Koeficienti snovnega

### Content (Syllabus outline):

- Basic definitions and concepts of process engineering.
- Heat Transfer. Mechanisms of heat transfer. Steady and unsteady heat transfer. Conductive, convective and radiation heat transfer. The coefficients of heat exchange. Correlations.
- Mass transfer. Mechanisms of mass transfer. Stationary diffusion. Convective mass transfer. The coefficients of mass

<p>prenosa. Korelacijske za izračun koeficientov prestopa.</p> <ul style="list-style-type: none"> <li>Separacijski procesi. Fazno ravnotežje. Kontinuirni in stopenjski kontakt v napravah. Načrtovanje in analiza separacijskih procesov. Sedimentacija. Filtracija. Centrifugiranje. Uparjanje. Kristalizacija. Destilacija. Ekstrakcija. Kromatografija. Dializa. Membranski procesi. Aplikacije v kemijski in farmacevtski industriji.</li> <li>Kemijska reakcija. Kinetični in termodinamski podatki. Kemijski reaktor. Hitrostna enačba. Enostavne in kompleksne reakcije. Kinetična analiza podatkov.</li> <li>Reaktorji za homogene reakcije. Šaržni reaktor. Mešalni reaktor. Cevni reaktor.</li> <li>Heterogene katalizirane reakcije. Mechanizem in hitrostna enačba katalitske reakcije. Transport snovi in toplotne preko površine katalizatorja in znotraj katalitskega zrna. Globalna hitrost reakcije.</li> <li>Reaktorji za heterogeno katalizirane reakcije. Reaktor s strnjениm slojem. Reaktor s fluidiziranim slojem. Reaktor z goščo. Kapalni reaktor. Membranski reaktor.</li> <li>Osnove dimenzioniranja reaktorjev. Snovna in toplotna bilanca. Tok tekočin v mešalnikih in reaktorjih. Avtomatska regulacija kemijskih reaktorjev.</li> <li>Okoljevarstveni vidiki procesnih industrij, preprečevanje onesnaževanja na izvoru in obdelava neogibnih emisij.</li> </ul>	<p>transfer. The correlation coefficients for calculating the mass transfer fluxes.</p> <ul style="list-style-type: none"> <li>Separation processes. The phase balance. Continuous and speed contact devices. Design and analysis of separation processes. Sedimentation. Filtration. Centrifugation. Evaporation. Crystallization. Distillation. Extraction. Chromatography. Dialysis. Membrane processes. Applications in the chemical and pharmaceutical industries.</li> <li>Chemical reaction. Kinetic and thermodynamic data. Chemical reactor. Rate equation. Simple and complex reactions. Kinetic analysis of the reaction data.</li> <li>Reactors for homogeneous reactions. Batch reactor. Mixing reactor. Tubular reactor.</li> <li>Heterogeneous catalyzed reactions. The mechanism and rate equation of catalytic reactions. Transport of mass and heat to the surface of the catalyst and within the catalyst grains. The global reaction rate.</li> <li>Reactors for heterogeneously catalyzed reactions. The reactor with a packed bed. The reactor is a fluidized bed. The reactor with the slurry. Dropping the reactor. Membrane reactor. Biochemical reactions and reactors.</li> <li>Basics of reactors design. Material and thermal balances. Fluid flow in the mixers and reactors. Automatic control of chemical reactors.</li> <li>Principles of sustainable production. Resource conservation. Environmental aspects of process industries. Pollution prevention and emission treatment.</li> </ul>
--	--

#### **Temeljni literatura in viri / Readings:**

1. Hriberšek, M. (2005) *Procesna tehnika*, 1.del. Maribor: Univerza v Mariboru, Fakulteta za strojništvo.
2. Holloway, M.D. et al. (2012) *Process Plant Equipment*. New York: John Wiley and Sons.
3. Seader, J.D., Henley, E.J., Roper, D.K. (2011) *Separation Process principles*. 3rd edition, New York: John Wiley and Sons.
4. Vogel, G. H. (2005). *Process Development*. Weinheim: Wiley-VCH Verlag.

5. Levenspiel, O. (2003). *Chemical Reaction Engineering*. 3rd edn., New York: John Wiley and Sons.
6. Perry, H. R., Green, D. (2008). *Chemical Engineer's Handbook*. 8. edn., New York: McGraw-Hill.
7. Schwister, K. (2000). *Taschenbuch der Verfahrenstechnik*. Leipzig: Carl Hanser Verlag.
8. Katoh, S., Yoshida, F. (2009). *Biochemical Engineering*. Weinheim: Wiley-VCH.
9. Fränzle, S., Markert, B., Wüschmann, S. (2012). *Introduction to Environmental Engineering*. Weinheim: Wiley-VCH.

#### Cilji in kompetence:

- Učna enota prispeva k razvoju naslednjih splošnih in predmetno-specifičnih kompetenc:
- Osnovni namen predmeta je študente seznaniti z osnovami toplotnega in snovnega transporta, ki se odvijata med potekom kemijske ali biokemijske reakcije, z osnovami dimenzioniranja kemijskih in biokemijskih reaktorjev ter z najvažnejšimi separacijskimi procesi, uporabljanih v kemijskih in procesnih industrijah. Ta znanja so potrebna pri skupnem delu s strokovnjaki drugih profilov (strojniških, elektrotehniških, računalniških...) pri razvoju novih produktov in tehnologij.

#### Objectives and competences:

- Learning Unit contributes to the development of generic and subject-specific competences:
- The main purpose of the course is to acquaint students with the fundamentals of heat and mass transfer, which take place in chemical or bio-chemical reactions, the basics of design and sizing of chemical and biochemical reactors and the most important separation processes used in chemical and process industries . This knowledge is needed to work together with experts from other profiles (mechanical, electrical, civil and computer engineering ) in the development of new products and technologies.

#### Predvideni študijski rezultati:

- Znanje in razumevanje: Pridobiti sposobnost in osnovna znanja, ki jih potrebujejo pri načrtovanju procesov s snovno pretvorbo in separacijskih procesov.
- *Uporaba*: Osvojiti metodološke prijeme/orodja za analizo in sintezo kompleksnih kemijsko-tehnoloških sistemov. Reflektirati vsebine z drugih strokovnih disciplin in jih povezati s pridobljenim znanjem. Spretnosti uporabe domače in tuje literature ter drugih virov, zbiranje in interpretiranje podatkov, sposobnost kritične strokovne analize, branje in izdelav tehnoloških načrtov ter ustnega in pisnega (vključno elektronskega) poročanja.

#### Intended learning outcomes:

- Knowledge and understanding: The ability to obtain and basic skills that students will need in process design with material conversion and separation processes.
- Application: Acquire the methodological techniques / tools for the analysis and synthesis of complex chemical- technological systems. To reflect the content of other professional disciplines and connect them with knowledge. Skills of using of domestic and foreign literature and other sources, acquisition and interpretation of data, critical analysis, reading and elaboration of technological plans, and oral and written (including electronic ) reporting.

**Metode poučevanja in učenja:**

- *predavanja z aktivno udeležbo študentov* (razlaga, diskusija, vprašanja in odgovori, teoretični in računski primeri, reševanje tipskih problemov – problemski pristop);
- *vaje in laboratorijske vaje* (načrtovanje in izvajanje poskusov, analiza eksperimentalnih podatkov, reševanje strokovnih problemov, projektno delo, dokumentiranje);
- *samostojno delo študentov* (individualni študij, priprava, predstavitev in zagovor projektne ali raziskovalne naloge).

**Learning and teaching methods:**

- lectures with active participation of students (explanation, questions and answers, discussion, theoretical and computational examples, solving of model problems - problem approach );
- exercises and laboratory work (design and conduct experiments , analyze experimental data, solving technical problems, project work , documentation);
- individual work of students ( individual study, preparation, presentation and defense of project or research paper).

Delež (v %) /

**Načini ocenjevanja:**

Weight (in %)

**Assessment:**

seminarska naloga	33 %	coursework
pisni izpit	33 %	written exam
ustno izpraševanje	34 %	oral exam